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feel on skin. 2. In the Arrhenius definition, an acid is a substance that produces hydrogen ions ( $H^+$ ) when dissolved in water, whereas a base is a substance that produces hydroxide ions ( $OH^-$ ) in aqueous solution.

### **Chapter 16 Acids and Bases**

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Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company ISBN 978-0-07802-152-7

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Using the techniques explained in Chapter 7, we can balance the equation.  $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2(g) + 4H_2O(g)$  Check: 3 C, 8 H, 10 O  $\rightarrow$  3 C, 8 H, 10 O This equation can be interpreted in terms of molecules as follows: 1 molecule of  $C_3H_8$ ! 5 molecules of  $O_2$   $\rightarrow$  3 molecules of  $CO_2$ ! 4 molecules of  $H_2O$

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12. 1 mol O = 16.00 g O =  $6.02 \times 10^{23}$  O atoms 1 O atom 16.00 g  
O  $6.022 \times 10^{23}$  O atoms = 2.66  $10^{-23}$  g O 13. 0.50 mol O atoms  
16.00 g O 1 mol = 8.0 g O 4 mol H atoms 1.008 g 1 mol = 4 g H  
Half a mole of O atoms weighs more than 4 moles of H atoms.

### **Chapter 6 Chemical Composition**

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